

# ***Multifunctional, Self-Healing Polyelectrolyte Gels for Long-Cycle-Life, High-Capacity Sulfur Cathodes in Li-S Batteries***

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Project ID  
bat320

# Overview

## Timeline

- Project start date – Oct. 2016
- Project end date – Dec. 2019
- Percent complete – 83%

## Budget

- Total project funding
  - DOE share: \$1.25 M
  - Contractor share: \$138,888
- Funding received in FY 2018  
\$416,667
- Funding received for FY 2019  
\$416,667

## Barriers

- **Cost:** Reduce \$/kWh of EV batteries using high-energy-density, low-cost Li-S chemistry
- **Performance:** Double the energy density of state-of-the-art Li-ion batteries using Li metal anode
- **Life:** Mitigate capacity loss mechanisms in Li-S cells for improved cycle life

## Partners

- Project Lead: University of Washington
- Interactions/collaborations: Pacific Northwest National Lab

# Relevance

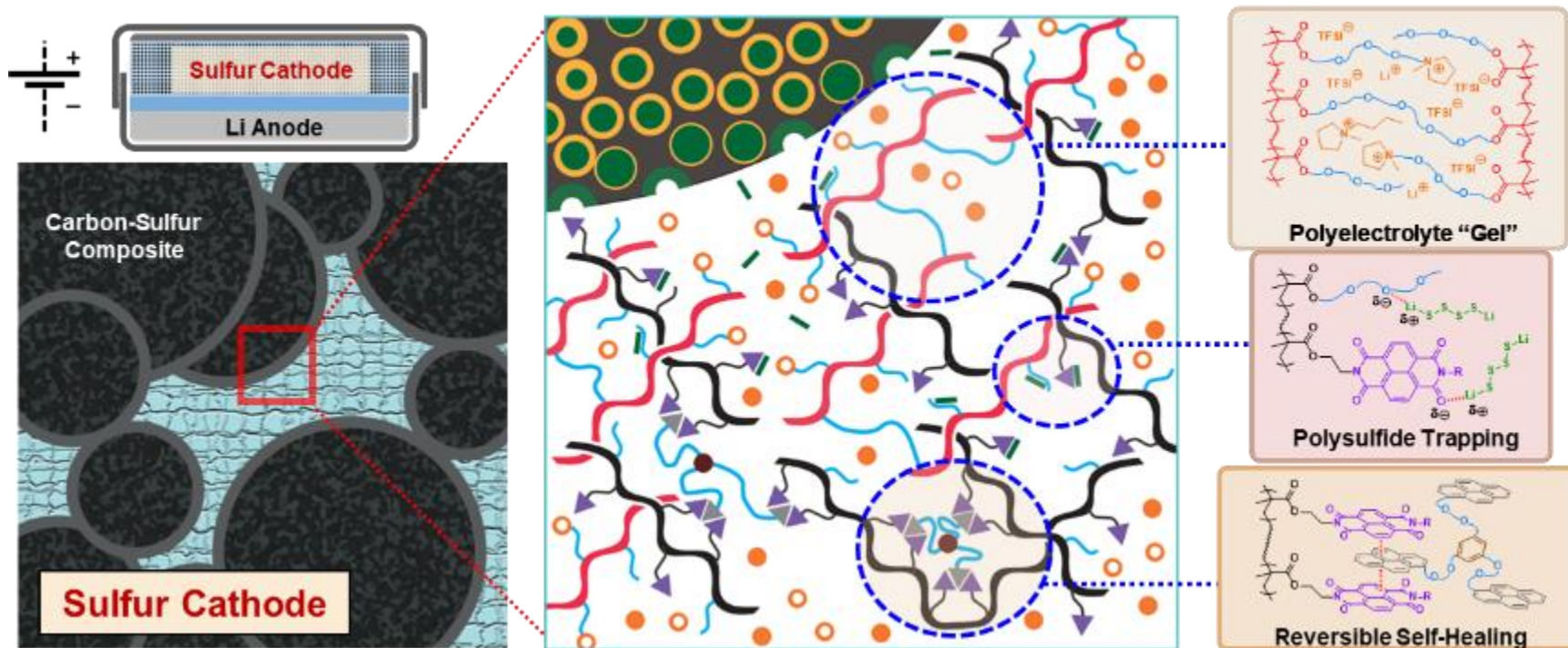
- **Overall Objective:** Develop high-performance Li-S cells, based on self-healing and polysulfide-trapping polyelectrolyte gels containing solvate ionic liquid (SIL). The Li-S cell design will be capable of achieving gravimetric energy densities in excess of commercial Li-ion cells (>250 Wh/kg).
- **Objectives This Period**
  - Develop *in-situ* fabrication methods for sulfur/carbon/ionogel cathodes and composite ionogel separators
  - Demonstrate quasi-solid-state (“all-gel” or QSS) Li-S cells and evaluate their behavior
  - Investigate the influence of self-healing, polysulfide-trapping materials on S cathode performance
- **Impact**
  - Li-S batteries have the potential to achieve the DOE goal of \$100/kWh for battery pack usable energy

# Milestones

Date	Milestone or Go/No-Go Decision	Status
	<u>Milestone</u>	
Sep. 2019	Optimized Cell Performance Update	On track
	<u>Milestone/Deliverable</u>	
Dec. 2019	Optimized Cell Performance Update / Deliver 10mAh Cells for DOE Testing	On track

# Approach/Strategy

## Multifunctional Polyelectrolyte Gels for Long-Cycle-Life, High-Capacity Li-S Batteries



# Approach/Strategy

- **Solvate Ionic Liquid + Polyelectrolyte Gel**
  - Ionic liquid electrolyte suppresses  $\text{Li}_2\text{S}_x$  dissolution and inhibits Li dendrite growth while providing conductivity similar to organic electrolytes
  - Solvate ionogel (SIG) made of cross-linked polyelectrolytes with solvate ionic liquid creates mechanical toughness without sacrificing ionic conductivity
- **Trapping of Polysulfide ( $\text{Li}_2\text{S}_x$ ) Species**
  - Containment of  $\text{Li}_2\text{S}_x$  species *via* physical and chemical interaction on cathode surfaces eliminates redox shuttle effect, improving capacity retention/efficiency
  - Incorporation of redox-active naphthalene diimide (NDI) group anchors  $\text{Li}_2\text{S}_x$  species and improves sulfur utilization by redox mediator effect
- **Self-Healing through Reversible Noncovalent Interactions**
  - Interaction of electron-rich pyrene (Py) group and electron-poor NDI group provides tunable/reversible self-healing, suppressing capacity loss due to mechanical degradation of cathode
- **Carbon/Sulfur Composite**
  - Mesoporous carbon provides conductivity and physical containment of  $\text{Li}_2\text{S}_x$
  - Platform adds targeted chemical functionality for performance enhancement

# Technical Accomplishments and Progress

## Highly-Conductive Solvate Ionogels (SIGs)



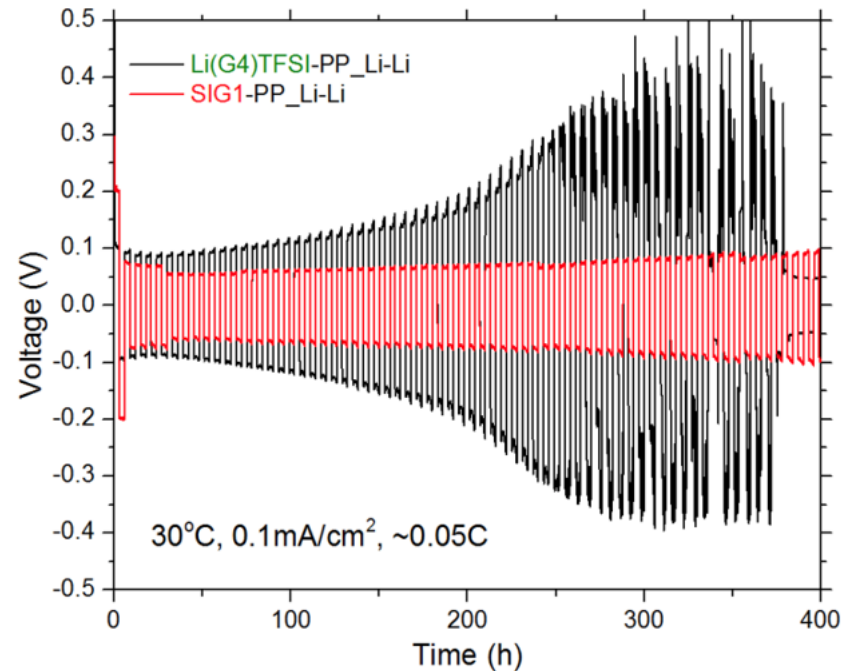
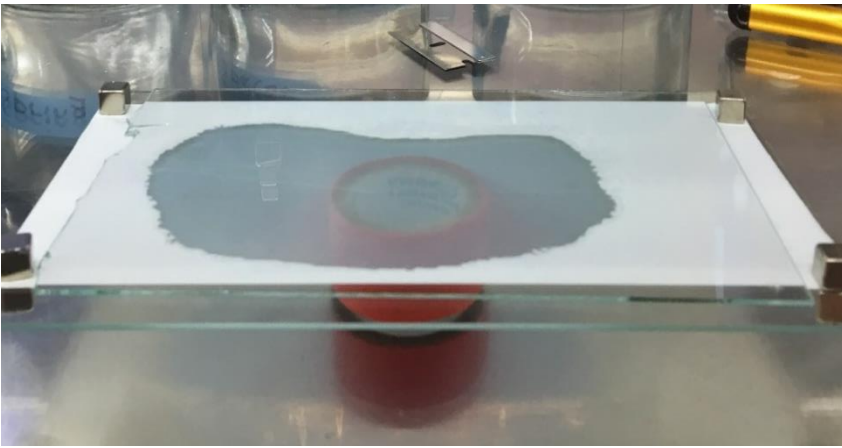
Formula	$\kappa$ @ 23°C (mS/cm)	$t_{\text{Li}^+}$ (EVB Method)
Li(G4)TFSI	1.08	0.13
“SIG 1”	0.73	0.21
“SIG 2”	1.05	0.28
“SIG 3”	0.92	0.24
“SIG 4”	1.07	0.16
“DSIG 5”	<b>2.15</b>	<b>0.57</b>

Note: This data has been previously presented and is included for context

- One-pot thermal cure of precursors can produce thin ionogel (SIG) films with remarkably high  $\text{Li}^+$  ion conductivity and transference number
- Simple design/fabrication allows SIG incorporation into separator and cathode with minimal processing

# Technical Accomplishments and Progress (cont.)

## SIG/Celgard Composite Separator



- Wetting/curing SIG resin into porous polypropylene (Celgard 2500) creates *composite ionogel separator*
- Gel composite outperforms solvate ionic liquid in Li|Li symmetric cells



# Technical Accomplishments and Progress (cont.)

## Quasi-Solid-State (“All-Gel”) Device Design

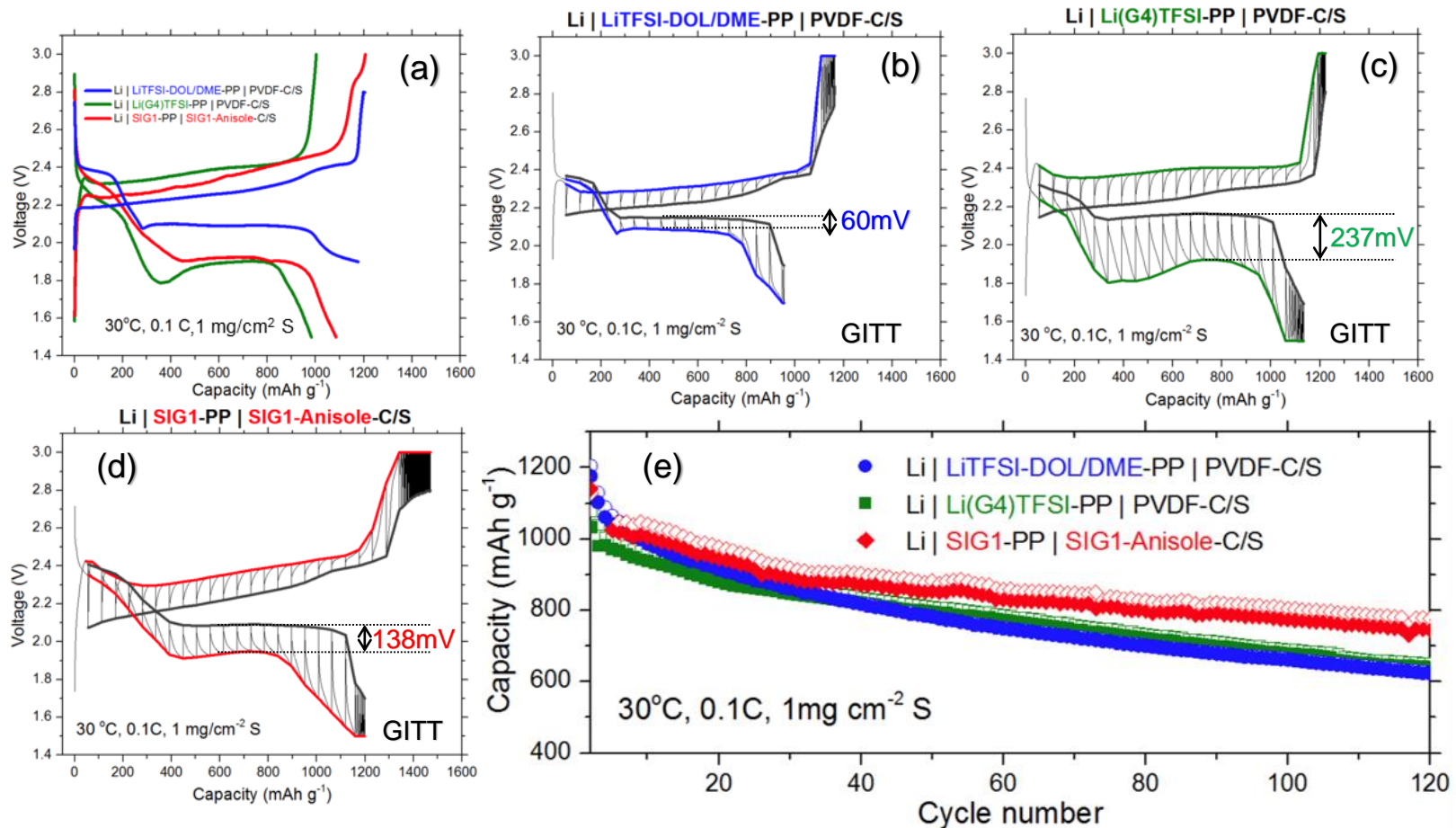


- C/S slurry made with SIG resin instead of PVDF binder and NMP solvent
- Slurry is cast and cured in place to produce an *in-situ* x-linked C/S/SIG cathode
- Gel cathode and separator laminated together with Li metal to produce a quasi-solid-state Li-S cell



# Technical Accomplishments and Progress (cont.)

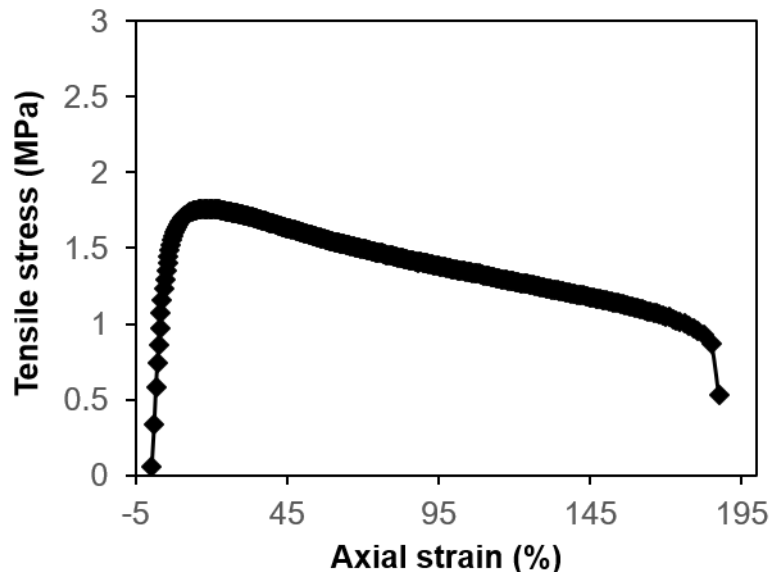
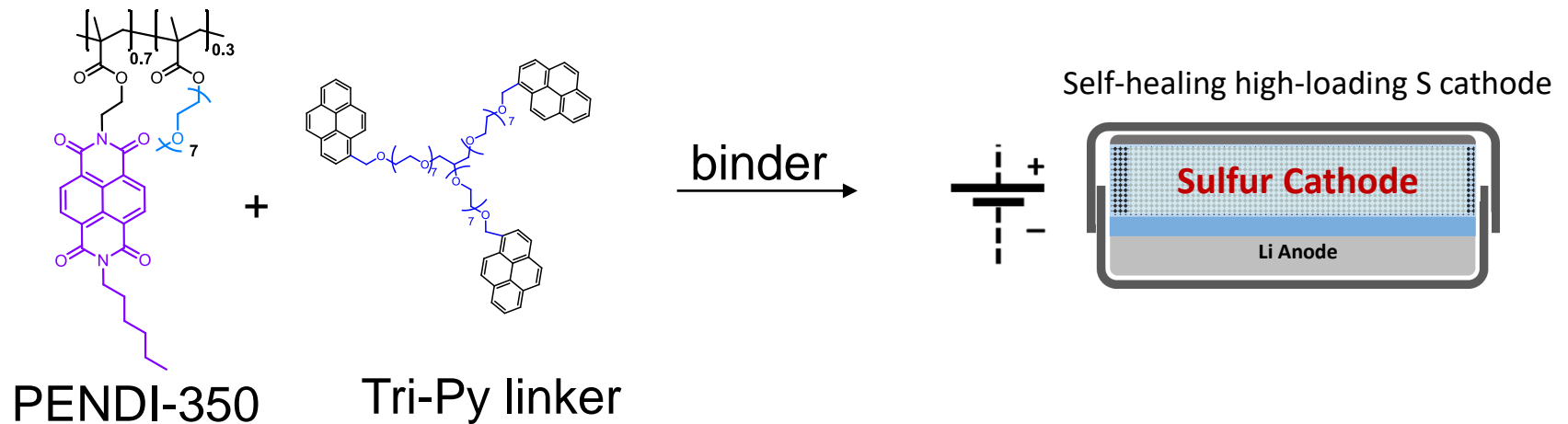
## Quasi-Solid-State Device Performance



- All-gel design gives improved capacity compared to organic electrolyte (w/LiNO<sub>3</sub>) or Li(G4)TFSI alone
- Overpotential reduced by using a “solvating diluent” anisole

# Technical Accomplishments and Progress (cont.)

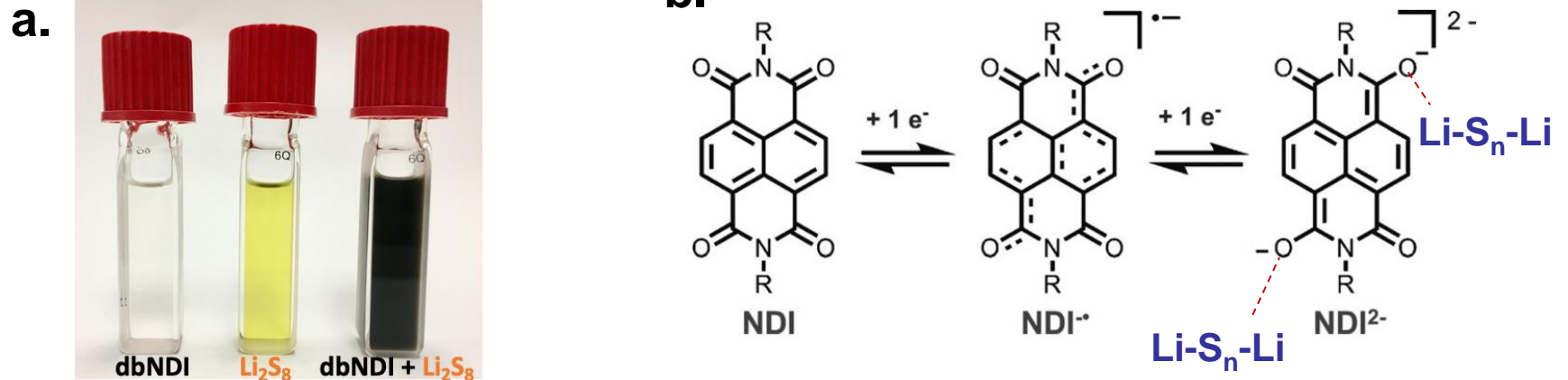
## Design of New Self-Healing Polymers



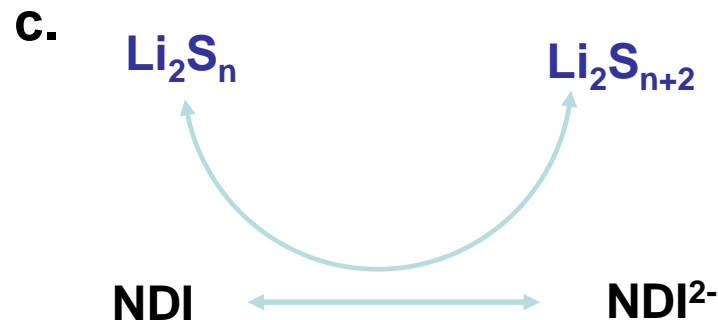
- Young's modulus: 29.2 MPa
- Maximum tensile stress: 1.8 MPa
- Maximum tensile strain: 185%
- Ionic conductivity:  $2.57 \times 10^{-7}$  S/cm (RT) (doped w/ LiTFSI, 1:20  $\text{Li}^+:\text{[EO]}$ )
- Self-healing temperature: 30 °C

# Technical Accomplishments and Progress (cont.)

## Interaction with $\text{Li}_2\text{S}_x$ and Redox-Mediator Effect of NDI



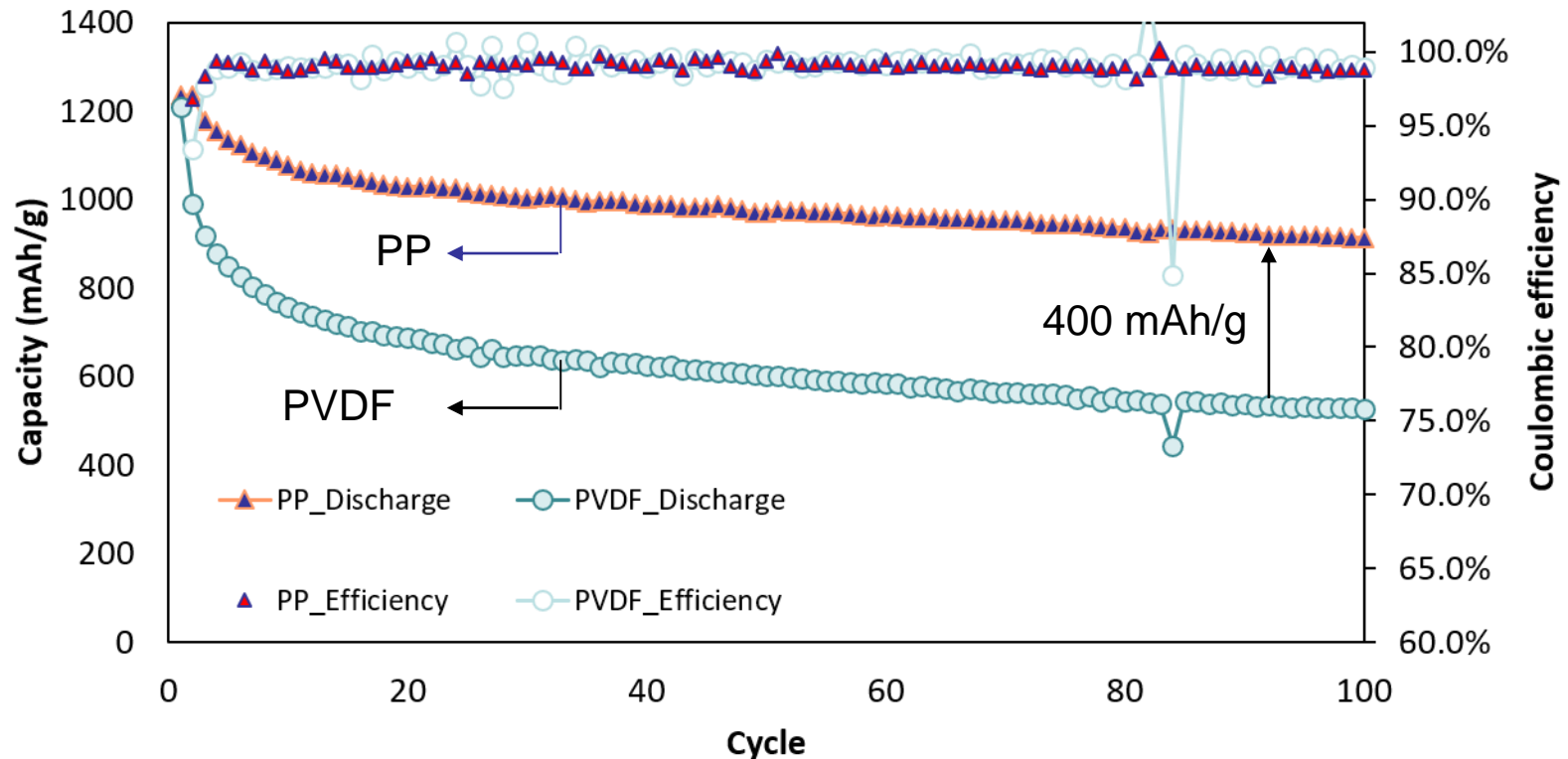
- Two-step reduction of NDI & ion-dipole interaction with  $\text{Li}_2\text{S}_x$



- $e^-$  exchange with NDI accelerates conversion of  $\text{Li}_2\text{S}_x$  (redox mediation effect)

# Technical Accomplishments and Progress (cont.)

## Performance of S Cathode w/Self-Healing Polymers

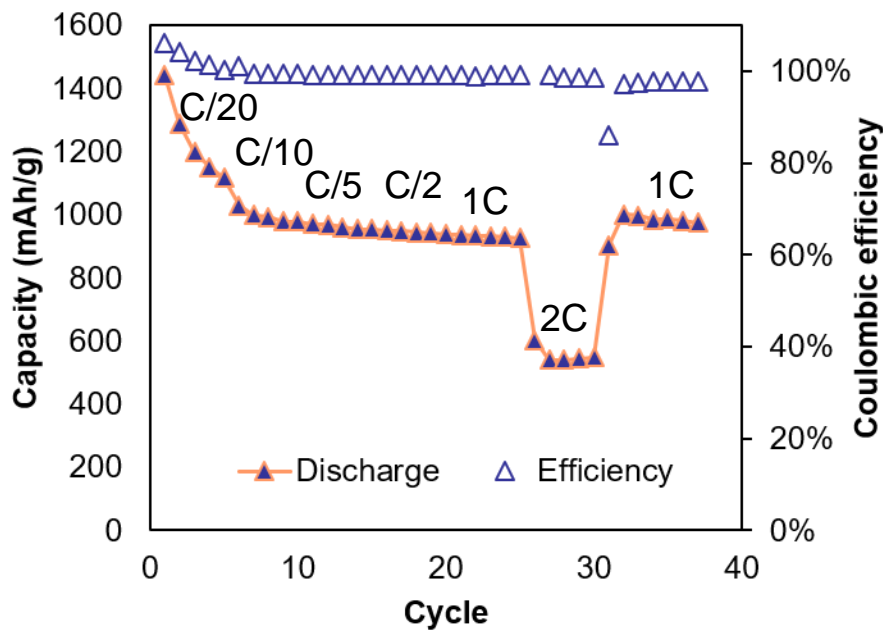


- Capacity with new self-healing PP polymer retains more than 87% (~910 mAh/g) after 100 cycles compare to control device using PVDF (75%, ~5300 mAh/g) at C/20 with S loading of ~0.9 mg<sub>S</sub>/cm<sup>2</sup>

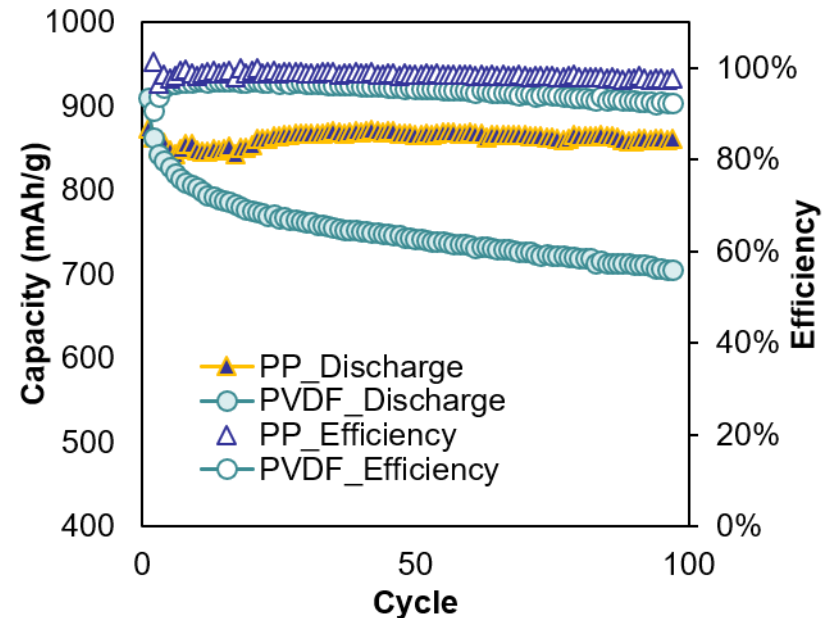
# Technical Accomplishments and Progress (cont.)

## Performance of S Cathode w/Self-Healing Polymers

**Different C rate**



**At 1 C rate**



- Capacity reaches about 980 mAh/g at 1C with S loading of  $\sim 1.0 \text{ mg}_S/\text{cm}^2$
- Capacity retains 98% after 100 cycles with S loading of  $\sim 1.1 \text{ mg}_S/\text{cm}^2$
- Testing of cathodes with  $> 4 \text{ mg}_S/\text{cm}^2$  is on-going

# Responses to Previous Year Reviewers' Comments

- This project was not reviewed at last year's AMR.

# Collaboration and Coordination with Other Institutions

- The project is carried out by the Jen and Yang groups at the University of Washington
  - Jen: design, characterize, and optimize ionogel electrolytes and self-healing materials, perform surface modification of mesoporous carbon, electrochemically characterize materials and concept cells
  - Yang: characterize carbon/sulfur composites and surface modified mesoporous carbon, design and electrochemically characterize practical cells
- FY 19 Collaborator – Pacific Northwest National Laboratory
  - Diffusion NMR characterization of ionogels



# Remaining Challenges and Barriers

- We have successfully produced high-loading ( $>4 \text{ mg}_\text{S}/\text{cm}^2$ ) cells, but must continue to optimize our fabrication methods and collect performance data.
- The primary degradation mechanisms in practical (high-loading) cells, as opposed to demonstration cells, must be identified and addressed.
- We must directly observe whether self-healing occurs during cathode operation with our novel binders, and identify its effect on cycle life
- We must identify, fabricate, and test the most efficient cell designs which integrate our approaches (quasi-solid-state cells, self-healing binder)

# Proposed Future Research

- **Before September 2019**

- Optimize the design and fabrication of high-loading ( $>4\text{mg}_\text{S}/\text{cm}^2$ ) S cathodes containing SIGs
- Adjust design of C/S/SIG cathodes to produce optimal conductivity and  $\text{Li}_2\text{S}_x$  solubility
- Demonstrate self-healing behavior in a functioning S cathode
- Efficiently integrate self-healing materials into quasi-solid-state design
- Identify prominent performance degradation pathways in Li-S cells with novel materials using a combination of microscopy, spectroscopy, and electrochemical behavior

- **Before December 2019**

- Fabricate optimized 10mAh coin cells based on our designs and deliver to designated DOE facility for validation testing

*Any proposed future work is subject to change based on funding levels.*

# Summary

- **Relevance**

- Rational molecular design of electrolyte materials has potential to systematically address Li-S cell performance issues, leading to a battery system with 2x energy density compared to Li-ion and high capacity retention

- **Approach**

- Develop freestanding gels containing solvate ionic liquid for Li metal compatibility and polysulfide solubility reduction
- Incorporate polysulfide trapping and redox mediation functions into polymeric binders using inter-molecular interactions to retain and effectively utilize sulfur in cathode
- Develop self-healing materials based on NDI/Py to heal mechanical damage during extended cell operation

- **Technical Progress**

- High conductivity demonstrated for a series of freestanding ionogel electrolytes
- Long-term compatibility of ionogels with Li metal demonstrated
- Ionogels seamlessly integrated into cathode and separator to produce quasi-solid-state Li-S cells
- Quasi-solid-state cells shown to cycle with improved capacity over traditional designs
- Significantly-improved cathode performance (capacity, retention, rate-performance) demonstrated using NDI/Py-based polymer binder
- Redox mediator effect of NDI contributes to improved sulfur utilization

- **Future Work**

- Optimize and characterize selected gel cathodes with improved cell performance
- Identify and develop methods to mitigate degradation pathways of our cell designs
- Deliver 10mAh optimized cells for DOE testing